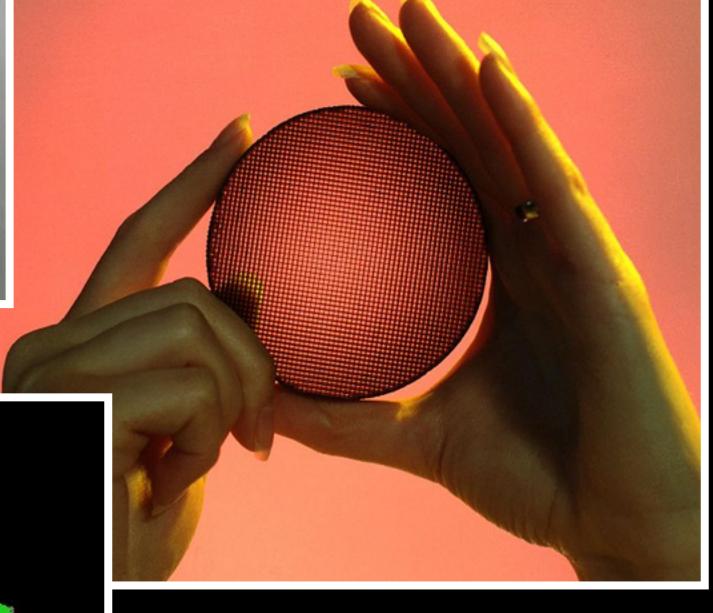
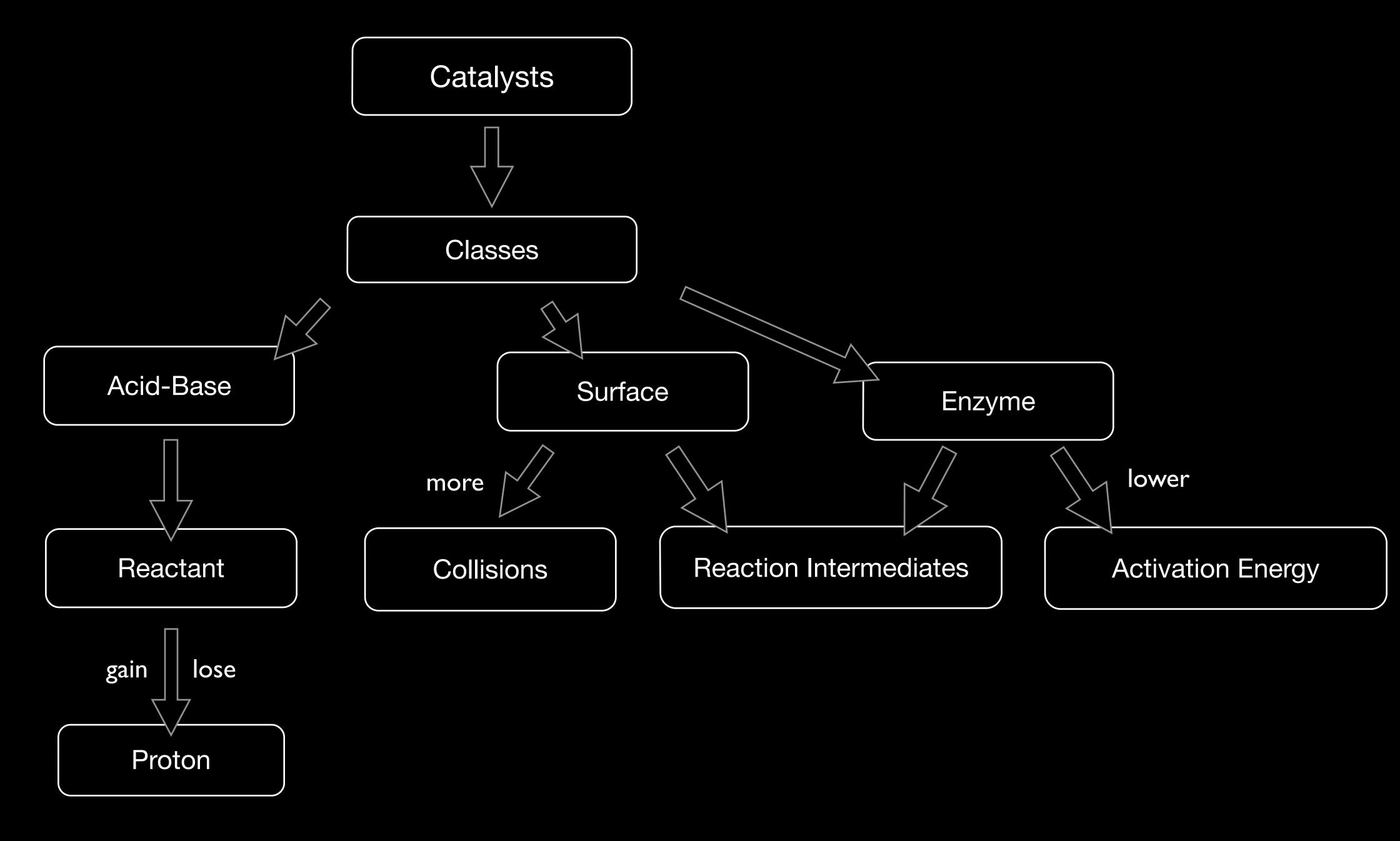


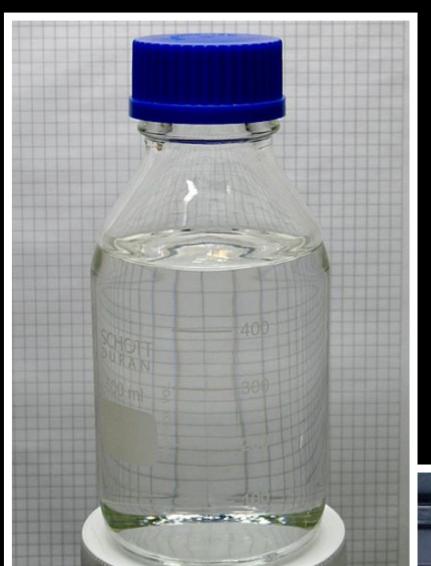
# Catalyst Classes

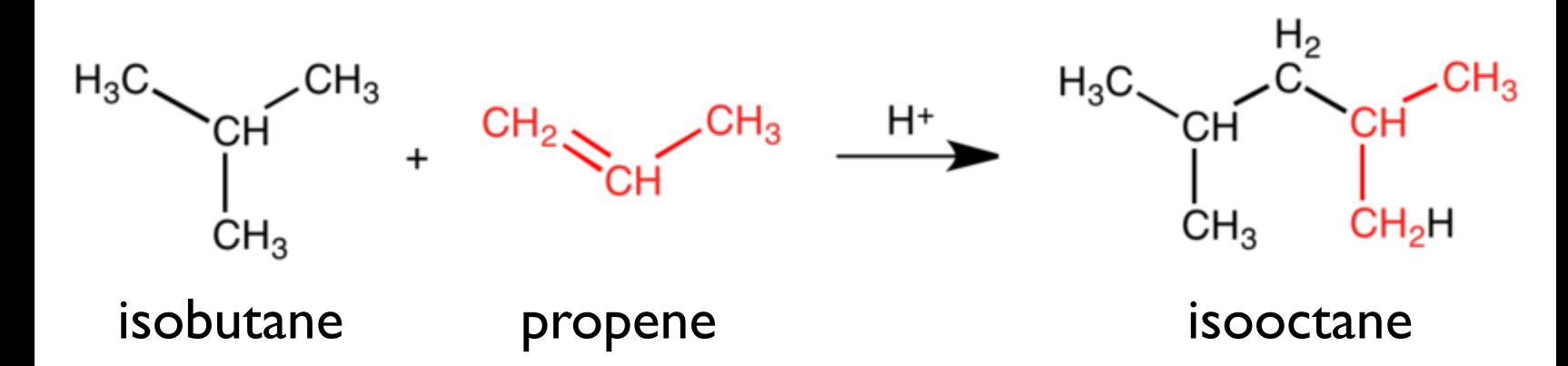


Chemistry Essentials - 045



#### Acid-Base Catalyst

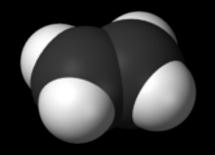




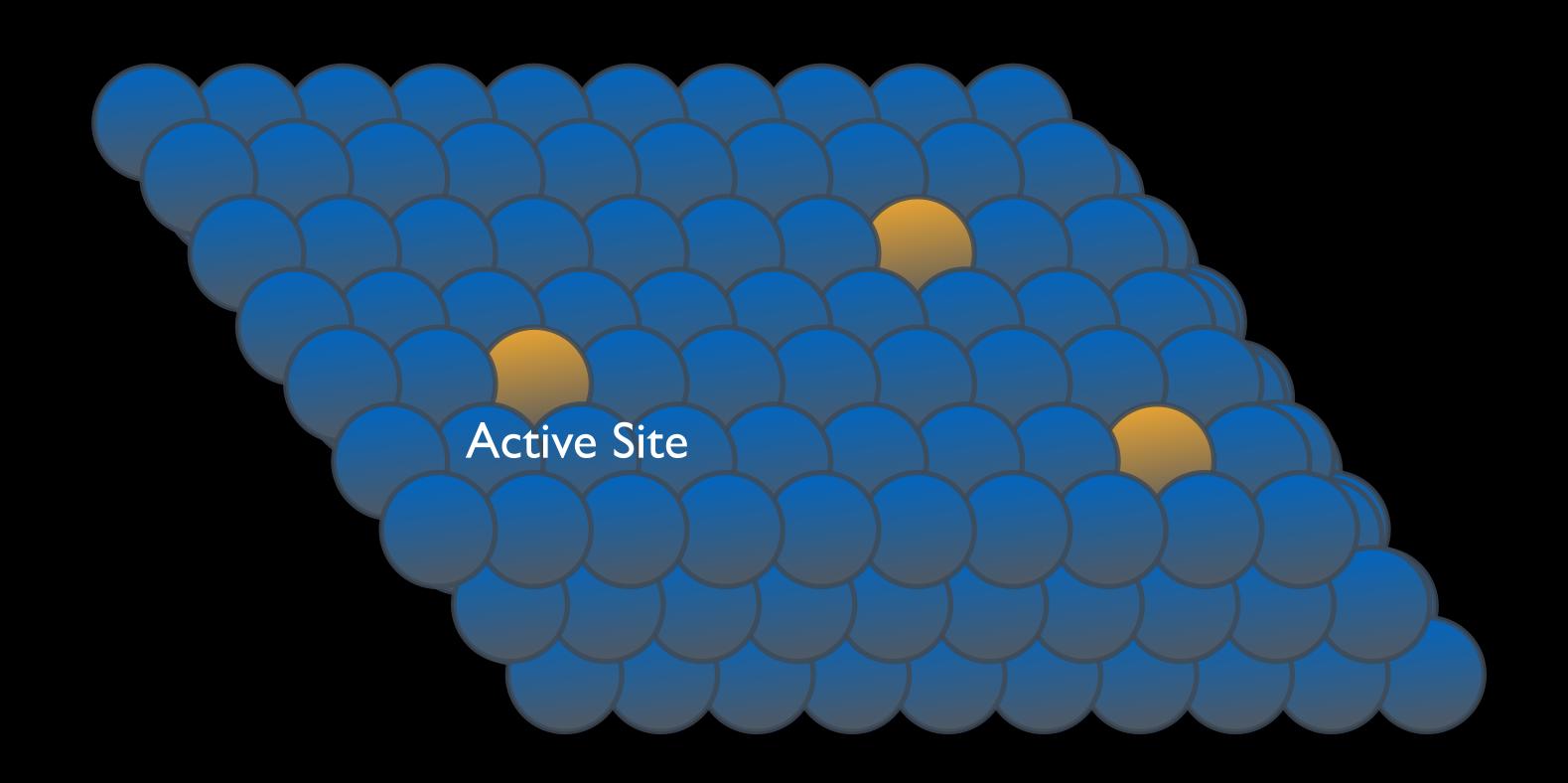


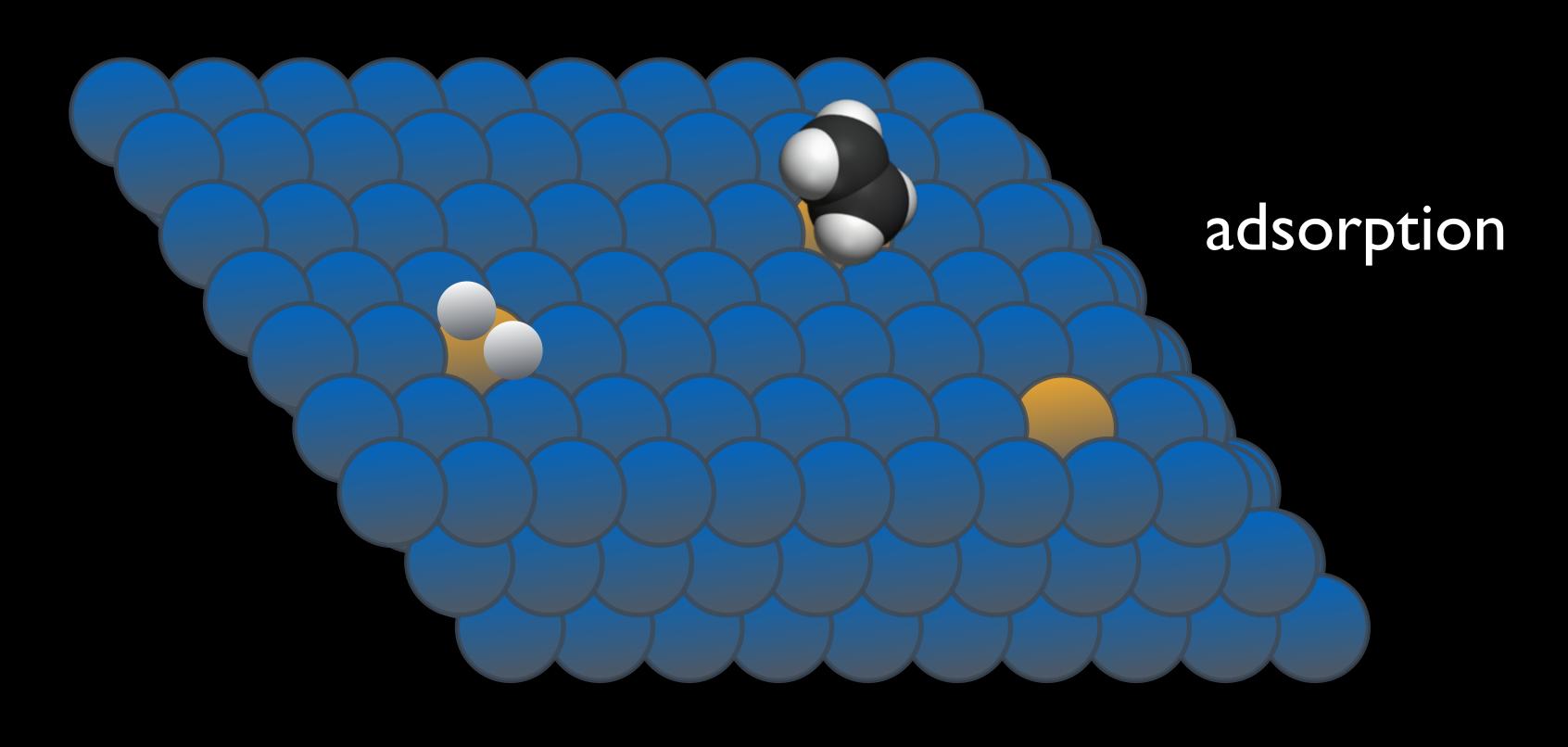
Hydrogen

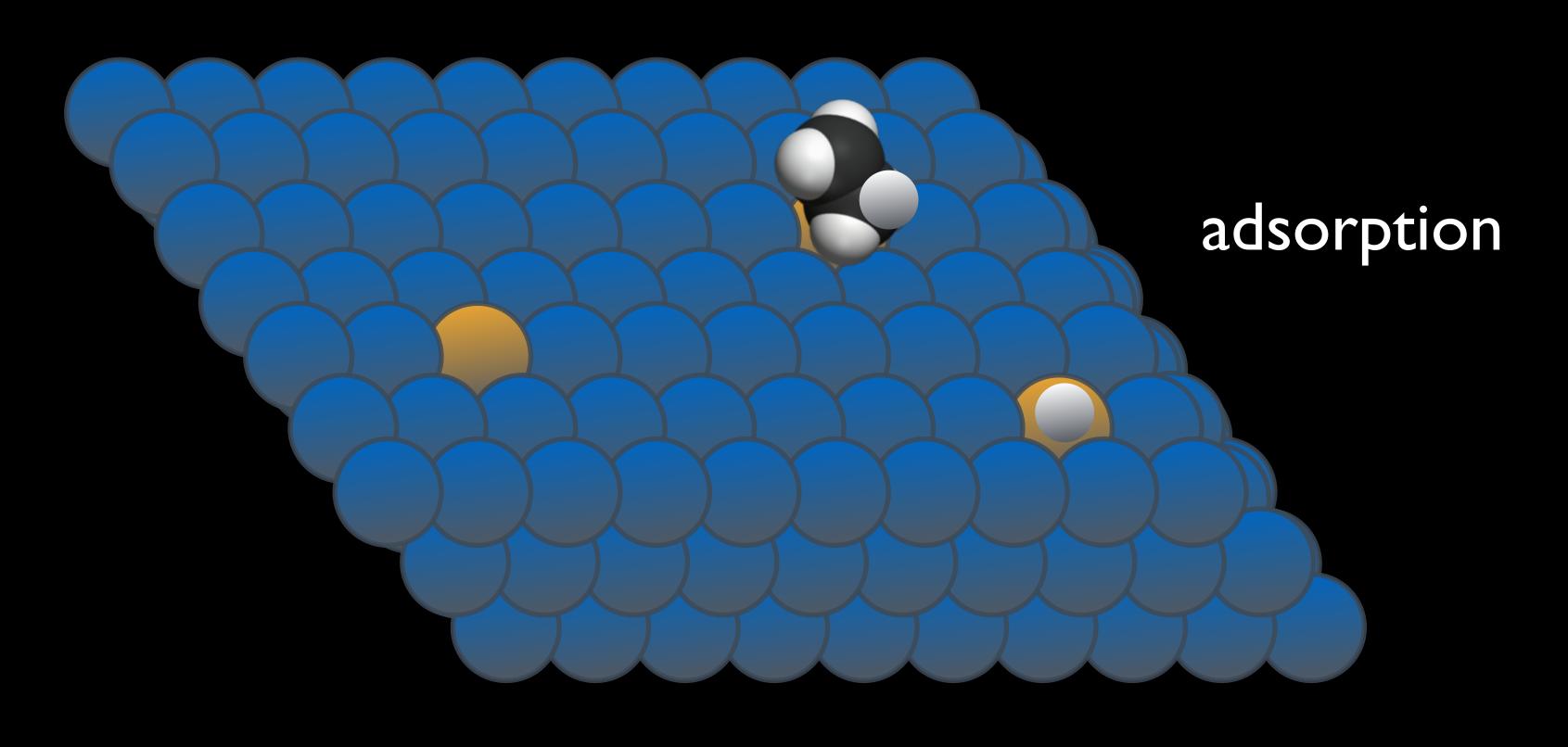




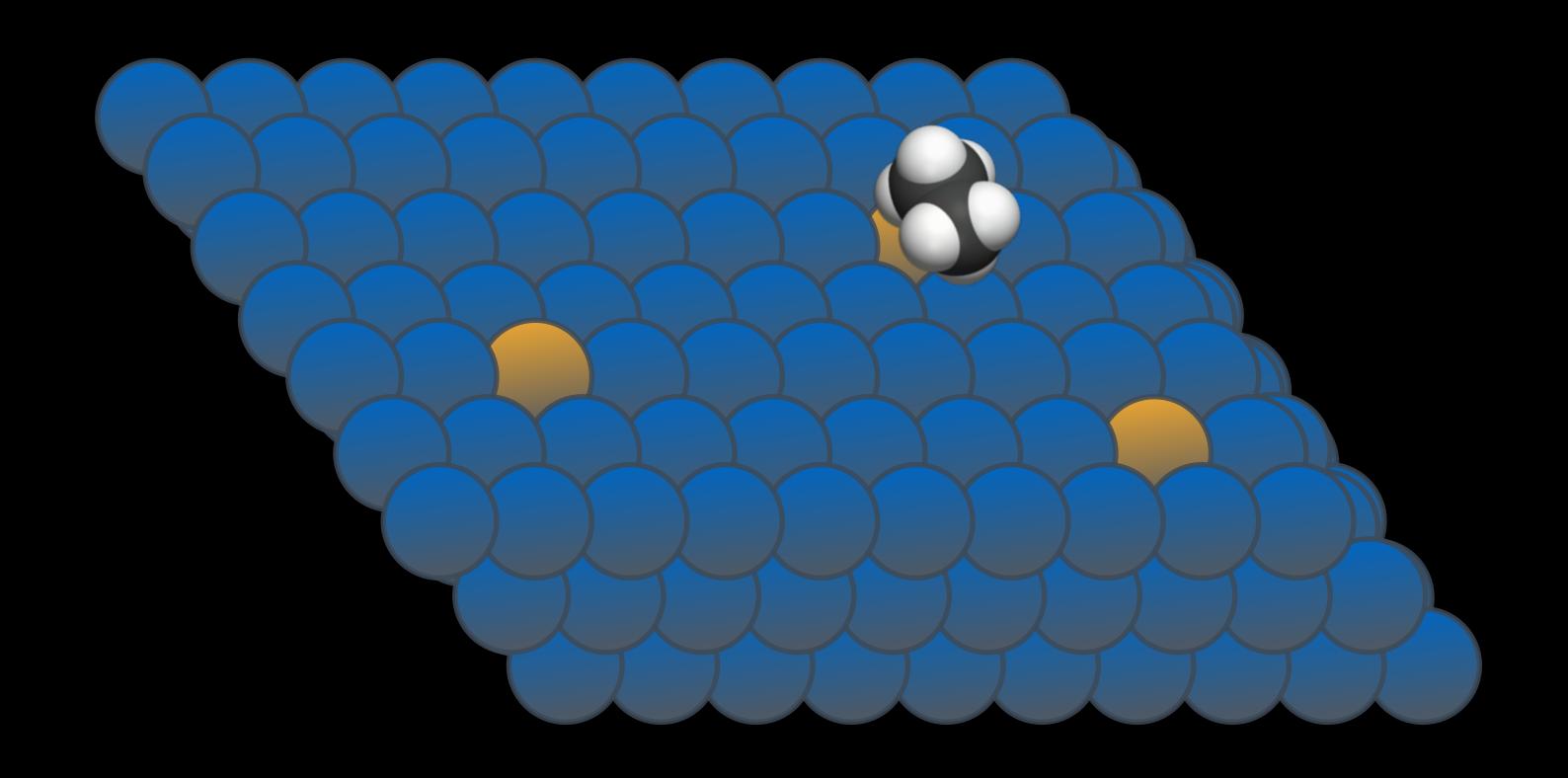
Ethylene

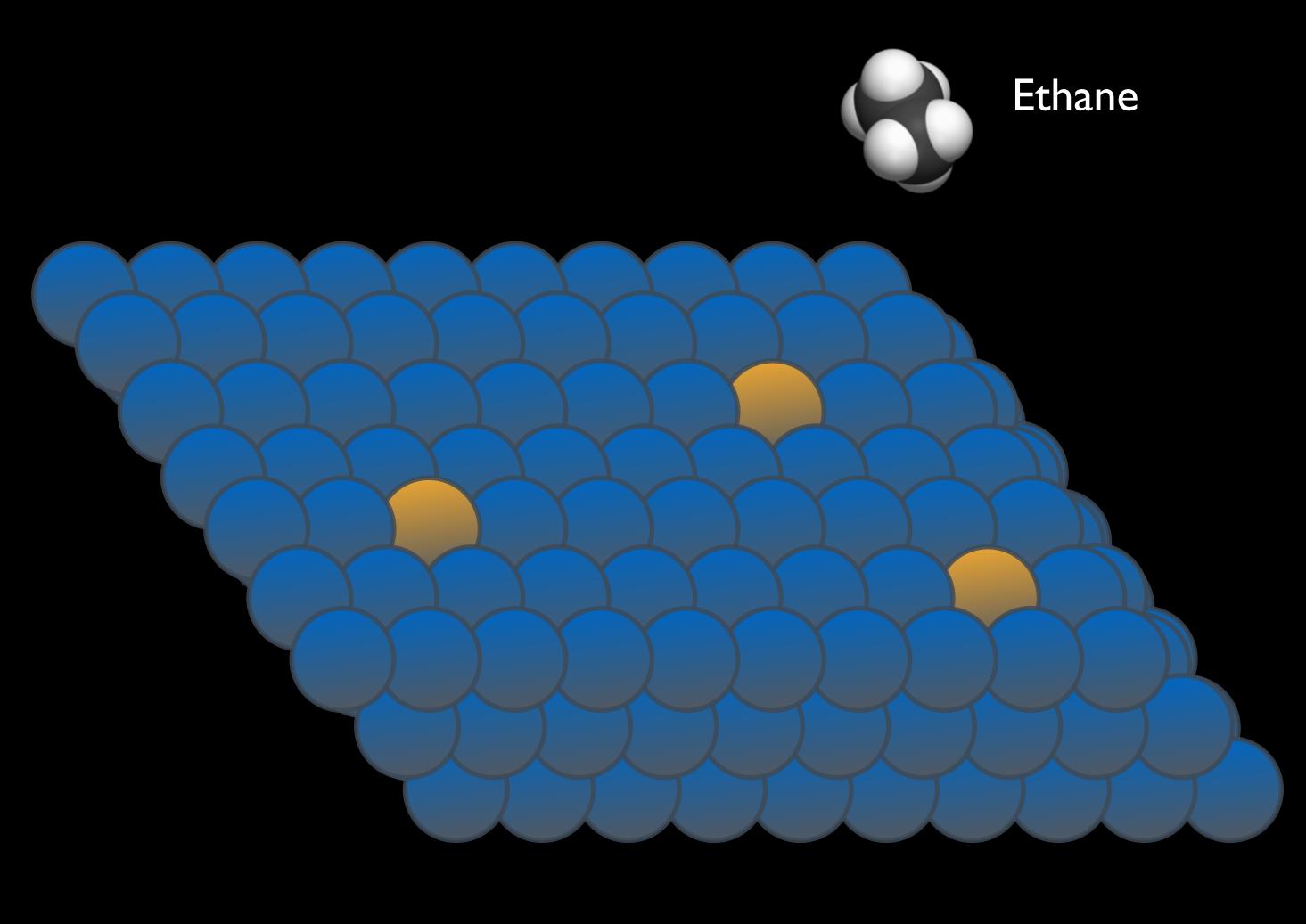


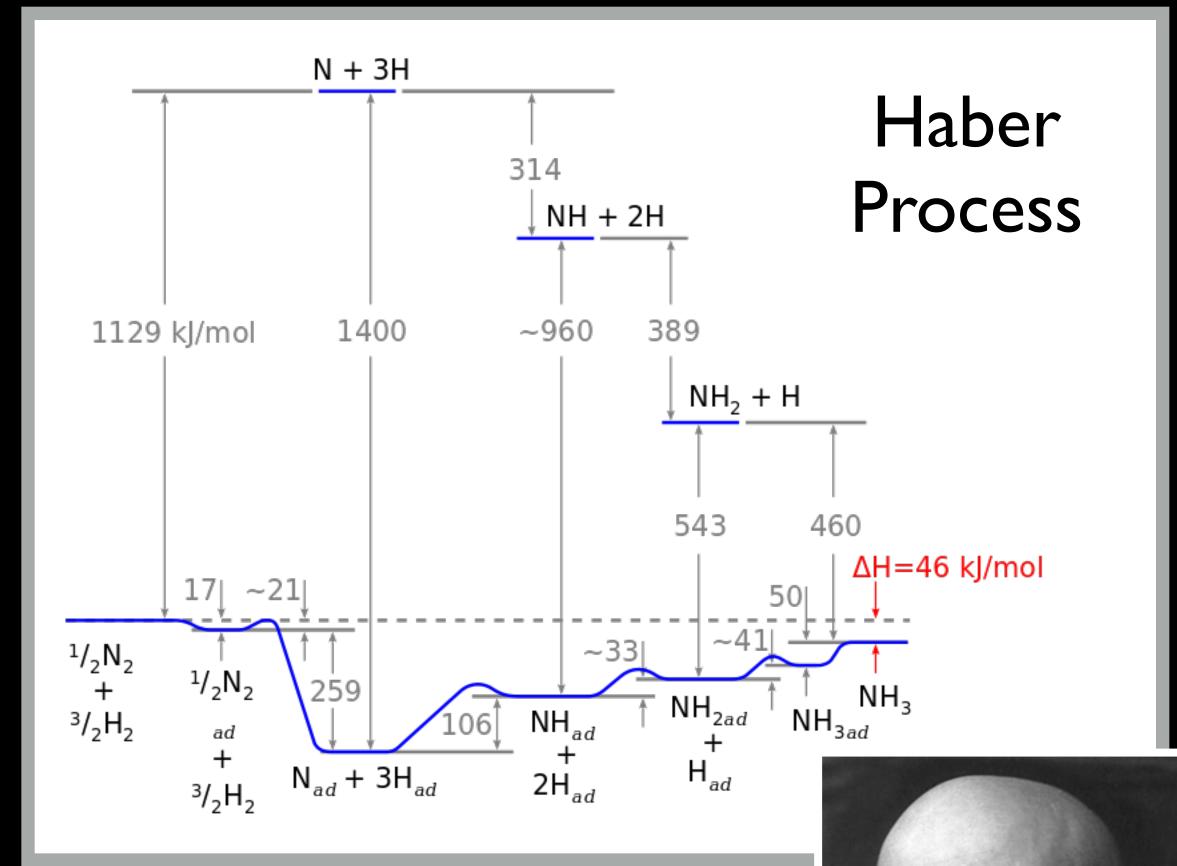


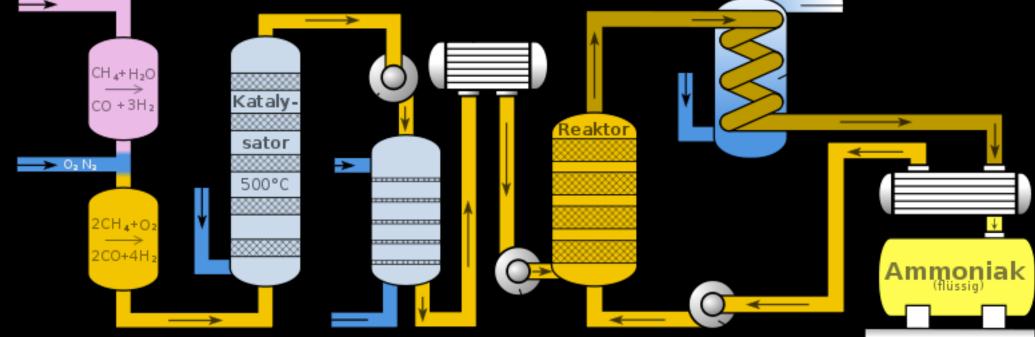


Ethane



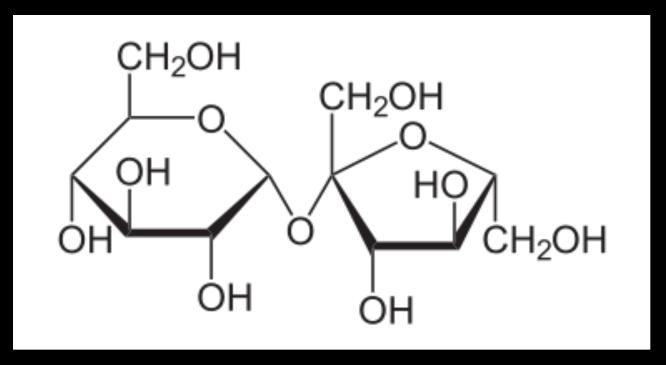






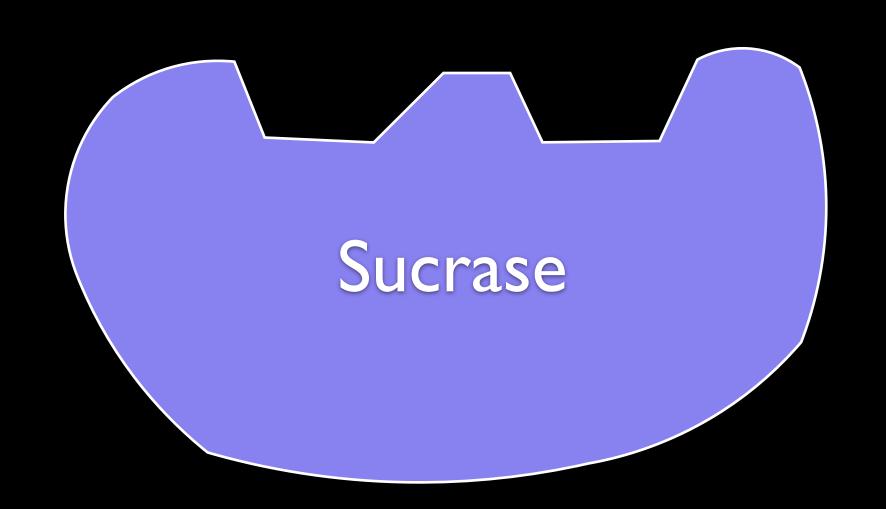
- 1.  $N_2(g) \rightarrow N_2$  (adsorbed)
- 2. N<sub>2</sub> (adsorbed) → 2 N (adsorbed)
- 3.  $H_2(g) \rightarrow H_2$  (adsorbed)
- 4. H<sub>2</sub> (adsorbed) → 2 H (adsorbed)
- 5. N (adsorbed) + 3 H(adsorbed) → NH<sub>3</sub> (adsorbed)
- 6.  $NH_3$  (adsorbed)  $\rightarrow NH_3$  (g)

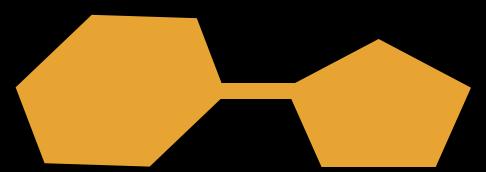




Sucrose





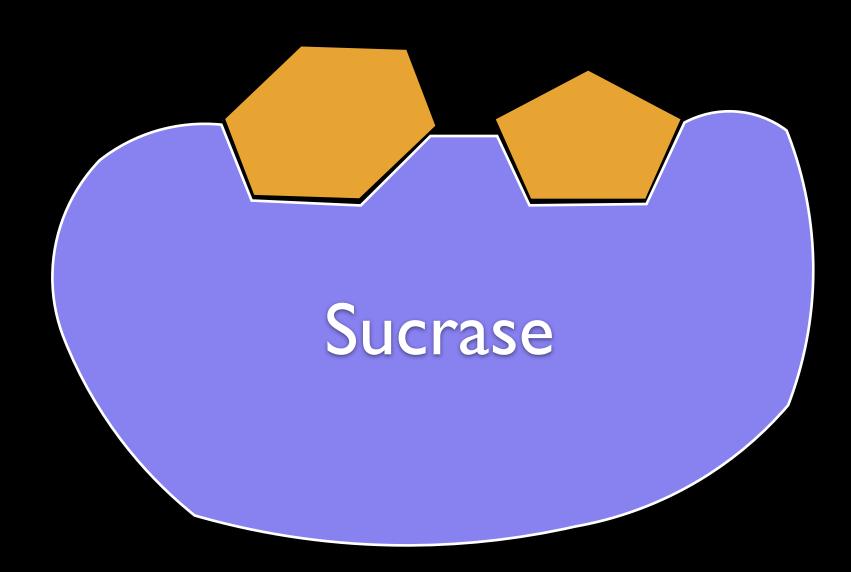


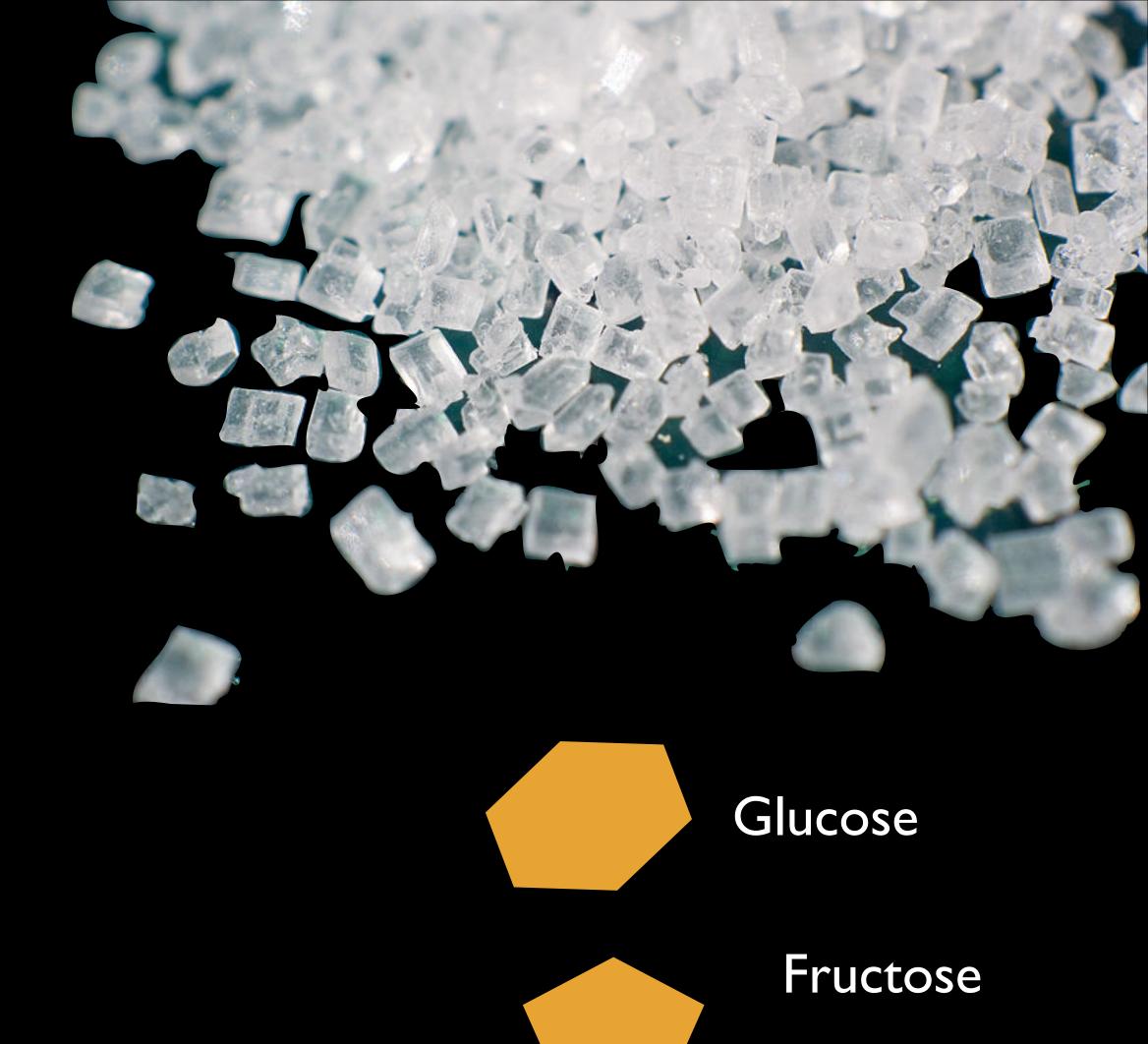
Sucrose

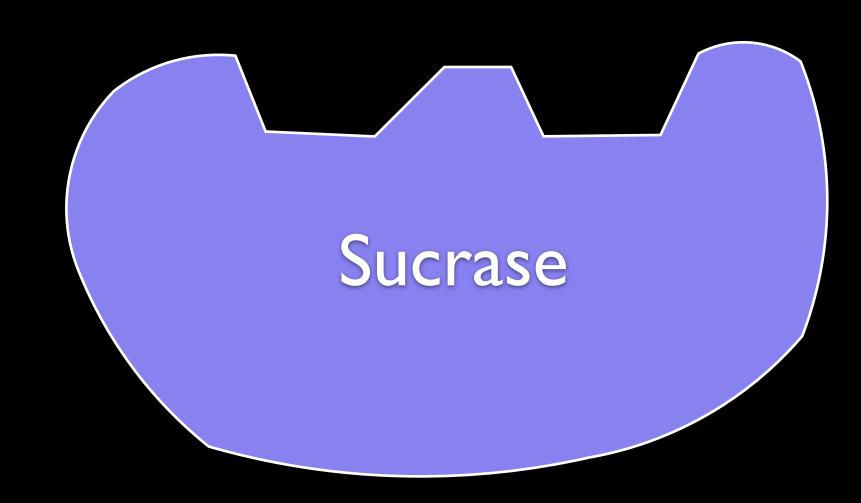




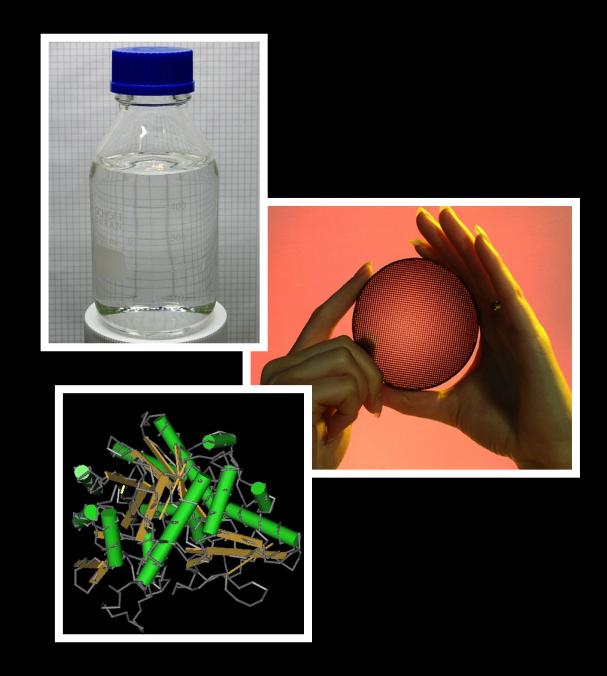
Fructose







#### Did you learn?



To explain changes in reactions arising from acid-base, surface and enzyme catalysts.

#### Acknowledgements

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